

Teaching Pack Pressure and the imploding can

Cambridge IGCSE™ Combined Science 0653

This *Teaching Pack* can also be used with the following syllabuses:

- Cambridge IGCSE[™] (9–1) Physics **0972**
- Cambridge IGCSE[™] (9–1) Co-ordinated Sciences (Double Award) 0973
- Cambridge IGCSE[™] Physical Science 0652
- Cambridge O Level Physics 5054





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Introduction

This pack will help you to develop your learners' experimental skills as defined by assessment objective 3 (AO3 Experimental skills and investigations) in the course syllabus.

Important note

Our *Teaching Packs* have been written by **classroom teachers** to help you deliver topics and skills that can be challenging. Use these materials to supplement your teaching and engage your learners. You can also use them to help you create lesson plans for other experiments.

This content is designed to give you and your learners the chance to explore practical skills. It is not intended as specific practice for Paper 5 (Practical Test) or Paper 6 (Alternative to the Practical Test).

There are two options for practising experimental skills. If you have laboratory facilities this pack will support you with the logistics of running the experiment. If you have limited access to experimental equipment and/or chemicals, this pack will help you to deliver a virtual experiment.

This is one of a range of *Teaching Packs*. Each pack is based on one experiment with a focus on specific experimental techniques. The packs can be used in any order to suit your teaching sequence.

The structure is as follows:



*the timings given here are guides; you may need to adapt the lessons to suit your circumstances.

In this *Teaching Pack* you will find the lesson plans, worksheets for learners and teacher resource sheets you will need to successfully complete this experiment.

Experiment: Pressure and the imploding can

This Teaching Pack focuses on pressure and the imploding can experiment.

Pressure is defined as the force per unit area. Atmospheric pressure, the pressure of the Earth's atmosphere around us, allows us to observe this concept in action. In this experiment, you will create low pressure inside the can and this will result in the atmospheric pressure imploding the can.

This experiment has links to the following syllabus content (see syllabus for detail):

- P1.6 Pressure
- P1.5 Effects of forces
- P3.1 Simple kinetic molecular model of matter

The experiment covers the following experimental skills, adapted from **AO3: Experimental** skills and investigations (see syllabus for assessment objectives):

- demonstrate knowledge of how to safely use techniques, apparatus and materials
- make and record observations.

Prior knowledge

Knowledge from the following syllabus topics is useful for this experiment.

- P1.5 Effects of forces
- P1.6 Pressure

Going forward

The knowledge and skills gained from this experiment will be useful for when you teach learners about the simple mercury barometer and the use of a manometer.

Briefing lesson: Pressure



	9
Resource Learning objective	 100 g mass and 2 drawing pins (for each learner) 2 pieces of white-tac for each learner Glass or plastic bottle and balloons Glass, cardboard (approx. 10 cm x 10 cm) and access to water Plunger and smooth surface (e.g. table or tile) Plastic bottle (or tin can) with two holes and access to water Worksheets A and B or C By the end of the lesson: all learners should be able to make observations. most learners should be able to record observations and identify the tools used to make them. some learners will be able to explain their observations in detail and link them to theoretical concepts.
Timinas	Activity
Things	Starter/Introduction
10 min	Encourage your learners to discuss and list their responses to the following questions: What tools can we use to make observations? How can we record our observations? What might we use our observations for? Collate your learners' ideas and discuss which are valid, efficient and most effective.
10 min	 Main lesson Learners should collect a 100 g mass (1N weight), two pieces of white-tac and two drawing pins. Explain to your learners how to carry out the experiment below: Make the two pieces of white-tac into slightly flattened balls which are large enough so they can put a drawing pin on top. Place one drawing pin on the first ball of white-tac with the point upwards. Place the 1N weight on top of the pin (they will need to support it). The learners should observe the force of the mass pressing the pin into the white-tac. Place the second drawing pin, point down, on the second ball of white-tac. They should not push the pin in, just hold it in position. Place the 1N weight on top of the effect. Invite learners to explain why the same weight is able to penetrate to a different depth in the white-tac depending on the orientation of the pin.
15 min 15 15 min	Provide your learners with <u>Worksheet A</u> . They should work in pairs to discuss the questions and write their own responses on the worksheet. Depending on timing, and your class, you could go through the answers orally or have them checked later. Your learners can work in pairs to carry out the four experiments and record their observations using <u>Worksheet B</u> . Abler learners can explain their observations. If
•••••	equipment cannot be provided, use <u>Worksheet C</u> . Plenary Return to the starter discussion. What tools have we used today? How have we recorded our observations? What have we used our observations for?

Lab lesson: Option 1 – run the experiment



Resource Learning objective	 Teacher notes Teacher walkthrough video Worksheets D and E Equipment as outlined in the notes By the end of the lesson: all learners should be able to make and record observations. most learners should be able to demonstrate knowledge of how to safely use techniques, apparatus and materials. some learners will be able to recognise that an imbalance of pressure can produce forces that change the size and shape of a body 	
	body.	
Timings	Activity	
10 min	Starter/Introduction Ask your learners to work in small groups to discuss and identify the necessary safety precautions for an experiment using a Bunsen burner and water. Collate their ideas through class discussion.	
10 min	 Main lesson Introduce the experiment and then demonstrate the setup. Make sure that you highlight the following safety points: Do not move the can until steam is clearly observed. Ensure you have a firm hold on the can before attempting to lift and move it. Check the area around you is clear before moving hot equipment. 	
20 min	Learners should collect the equipment and set up the experiment following your example. They should follow the method on <u>Worksheet D</u> .	
e CI min e.e.e	break down the process and explain it in detail. Less able learners may use Worksheet E to structure their explanation. Safety Circulate the classroom at all times during the experiment so that you can make sure that your learners are safe.	
	Plenary	
5 min	Return to the starter discussion from the briefing lesson. What tools have we used today? How have we recorded our observations? What have we used our observations for? Learners should discuss this in small groups before sharing their	

answers with the class.

Teacher notes



Watch the Teacher walkthrough video and read these notes.

Each group will require:

- access to water
- large tray
- ice (enough to make two ice-water baths in large trays)
- Bunsen burner, gauze and tripod
- heat mat
- can
- heat-resistant gloves OR tongs large enough to hold the can
- small beaker to measure 25 ml of water

Safety

The information in the table below is a summary of the key points you should consider before undertaking this experiment with your learners.

It is your responsibility to carry out an appropriate risk assessment for this experiment.

Substance	Hazard	First aid
_	Burns	Flood burnt area with water for at least 10 minutes. For serious injuries see a doctor.

Teacher method

This is your version of the method for this experiment that accompanies the *Teacher walkthrough* video.

Do not share this method with learners. Give them Worksheet D.

Before you begin

Plan how you will group your learners during the experiment session.

Think about:

- the number of groups you will need (group size 2–4 learners)
- the amount of equipment required
- whether you are demonstrating this experiment or allowing learners to attempt the experiment

Experiment

Circulate during the experiment in case learners encounter any difficulties.

Depending on the ability of the learners, you may wish to run this experiment as a demonstration instead. Ensure you have completed the experiment yourself before instructing the learners to attempt it.

Notes

Steps

- 1. Learners collect equipment from the front of the class.
- 2. They should prepare an ice-water______ bath in the large tray.
- 3. The Bunsen burner is placed on the heat mat. Over this, learners should arrange the tripod and gauze.
- 4. Learners should place the can on top of the gauze. They should practise either:
 - Using the heat-resistant gloves to remove the can, secure the lid and place in the tray.
 - b. Using the tongs to remove the can, then tip it over quickly so that the open end of the can is submerged in the ice-water.

Learners may share trays, but the learners' equipment must be close to the tray and they must communicate with each other to avoid both using it at the same time.

In the video a can with a lid is used (option **a**). The same experiment can be carried out with a ring-pull drinks can (option **b**). Decide in advance which method is to be used and advise your learners accordingly.

- Learners place the can on the desk and pour approximately 25 cm³ of water from the beaker into the can.
- Learners should place the can on top of the gauze and ensure that it is stable.
- Learners check that they are happy with the set-up and that everything is secure – does it match your demonstration of the set-up?
- Learners put on safety goggles and ensure chairs are out of the way. They must conduct this experiment standing up.
- 9. Learners light the Bunsen burner and ensure it is beneath the gauze and the can.
- 10. Learners observe the can.
- 11. When steam can be seen rising consistently from the can, it is time to perform the next stage of the experiment. Learners turn off the Bunsen burner. Either:
 - a. Using the heat resistant gloves to remove the can, secure the lid and place in the tray.
 - b. Using the tongs to remove the can, tip over quickly so that the ____ open end of the can is submerged in the ice-water.
- 12. Learners observe what happens next and record their observations from the experiment.

Clean-up

After the experiment learners should:

• Leave equipment to cool and then return the equipment to you.

It is better to have slightly less water than too much. With more water it will take longer to boil. For a smaller can, less water is advised.

Remind learners to check the stability of the equipment so there is no risk of boiling water being spilled during the experiment.

Waiting for the steam may take some time. Ensure learners remain focused on the task.

If this is not done quickly, the steam will condense, and the air pressure in the can will equalise.

Lab lesson: Option 2 – virtual experiment



Ask your learners to work in small groups to discuss and identify the necessary safety 10 precautions for an experiment using a Bunsen burner and water. min Collate their ideas through class discussion. Main lesson Provide your learners with Worksheet D. Ask them to read the method carefully and 10 identify the specific safety concerns for the experiment. Discuss these as a class. min Watch the video for the first time. Ask your learners to make initial notes of their observations. After the video discuss these as a class. Watch the video again, pausing to give extra time for learners to make more detailed observations as a class. Your learners should add to their previous observations. Run through the video a final time to discuss pressure and its effects in the experiment. Learners need to make and record observations. They can use their notes from watching the virtual experiment video and draw diagrams to break down the process and explain it in detail. Less able learners may use Worksheet E to structure their work. Plenary Return to the starter discussion from the briefing lesson. What tools have we used today? How have we recorded our observations? What have we used our observations for? Learners should discuss their responses in small groups before sharing their answers with the class.

Debriefing lesson: Understanding pressure

Resources	 Worksheets F, G, H and I Aerosol can
Learning objectives	 By the end of the lesson: all learners should be able to explain that an imbalance of pressure can produce forces. most learners should be able to explain that these forces change the size and shape of a body. some learners will be able to identify other examples of imbalances of pressure producing forces.

Timings	Activity	
	Starter/Introduction	
10 min	Provide your learners with <u>Worksheet F</u> . They should work through the worksheet independently to begin with. After 3 minutes, they should share their thoughts with a partner. You should then take feedback from each pair so that the whole class can share their ideas (think, pair, share). All learners should have the same answer to the first question, but answers to the others may differ. This is not a problem, as long as they can justify their thinking.	
	Main lesson	
15 min	Give your learners <u>Worksheet G</u> . They should work through the activities independently. You might want to discuss their ideas for the extension question. After sufficient working time, go through questions 1-3 as a class.	
15 ^{min}	Provide learners with <u>Worksheet H</u> . They can work through the worksheet independently or you may want to allow them to discuss and draft their letter in pairs. Circulate around the room to read what the learners are writing. Ask a few learners to read their letters to the class. Learners provide feedback to their peers about what is good in the letters and what could be improved.	
10 min	Show the class the aerosol can. Explain how an aerosol can contain a compressed fluid. When sprayed, the fluid is forced out of a small hole and spreads out. Aerosols are very dangerous when heated. Ask learners to discuss in small groups why you should never put an aerosol on a fire. Circulate around the room to hear ideas and direct discussion. Bring their ideas together as a class. You can find videos online to show what can happen if an aerosol is placed in a fire.	
	Plenary	
10 min	Provide learners with <u>Worksheet I</u> . They can work in pairs to match up the questions and answers. If there is time, they can cut out and stick the questions and answers in the correct order	

Worksheets and answers

	Worksheets	Suggested answers
For use in the <i>Briefing lesson</i> :		
A: Pressure	14	25
B: Atmospheric pressure	15–16	26–27
C: Atmospheric pressure	17–18	-
For use in <i>Lab lesson: Option 1</i> :		
D: Method	19	_
E: Observations of the imploding can	20	-
For use in <i>Lab lesson: Option 2</i> :		
D: Method	19	-
E: Observations of the imploding can	20	-
For use in the <i>Debriefing lesson</i> :		
F: Making observations	21	28
G: Calculating pressure	22	29
H: Taking precautions	23	-
I: Match up	24	30

Worksheet A: Pressure

Read through the information below before completing the questions in the space provided.

Pressure is defined as the force per unit area. The more concentrated the force over a smaller area, the higher the pressure.



Pressure is measured in Pascals (Pa), force is measured in Newtons (N) and area in metres squared (m²).

- 1. Fill in the spaces below:
 - **a.** 1 Pascal = N/m²
 - **b.** For a given area, increasing the force will the pressure.
 - c. For a given force, increasing the area will the pressure.
- 2. Discuss and answer the following questions.
 - a. Why do tractors have large flat tyres?









Worksheet B: Atmospheric pressure



Atmospheric pressure

- is caused by the weight of air above us
- decreases with altitude
- acts in all directions
- Average sea-level pressure is 101 000 Pa.

Write observations for the experiments below:

	What do you observe?
seal over the rim. Try to blow it up.	
	Explain why this happens.
Almost fill a class with water, place a square of	What do you observe?
cardboard firmly over the top and turn it upside	
down slowly.	
Carefully take your hand away and see what	
happens.	
	Explain why this happens.

Worksheet B: Atmospheric pressure



	What do you observe?
Press the plunger onto a smooth surface. Try to pull it away.	, ,
	Explain why this happens.
4	
	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
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The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the top hole is covered, what happens to the water coming out of the lower hole?	What do you observe?

Worksheet C: Atmospheric pressure



Atmospheric pressure

- is caused by the weight of air above us
- decreases with altitude
- acts in all directions
- Average sea-level pressure is 101 000 Pa.

Think about the role of atmospheric pressure in the experiments below:

A balloon is placed inside a bottle creating a seal over the rim. It is almost impossible to blow up.	Explain why this happens.
When a glass is almost filled with water and a square of cardboard is placed firmly over the top, it can be turned upside down slowly and the water stays in place.	Explain why this happens.
Carefully take your hand away and see what happens.	

Worksheet C: Atmospheric pressure



Press the plunger onto a smooth surface. Try to pull it away	What do you observe?
	Explain why this happens.
1	
	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the lid.	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the lid. When the top hole is uncovered, water comes out of the lower hole. When the top hole is	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the lid. When the top hole is uncovered, water comes out of the lower hole. When the top hole is covered, the water stops coming out of the	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the lid. When the top hole is uncovered, water comes out of the lower hole. When the top hole is covered, the water stops coming out of the bottom hole.	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the lid. When the top hole is uncovered, water comes out of the lower hole. When the top hole is covered, the water stops coming out of the bottom hole.	What do you observe?
The bottle has two holes, one higher than the other. It is filled with water and closed with the lid. When the top hole is uncovered, water comes out of the lower hole. When the top hole is covered, the water stops coming out of the bottom hole.	What do you observe?
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The bottle has two holes, one higher than the other. It is filled with water and closed with the lid. When the top hole is uncovered, water comes out of the lower hole. When the top hole is covered, the water stops coming out of the bottom hole.	What do you observe?

Worksheet D: Method



Follow the instructions to carry out the experiment.

- 1. Collect all of your equipment from the front of the class. You will need:
 - access to water
 - a large tray
 - ice (enough to make an ice-water bath in the large tray)
 - Bunsen burner, gauze and tripod
 - heat mat
 - an empty can
 - heat-resistant gloves OR tongs large enough to hold the can
 - small beaker to measure 25 cm³ of water
- 2. Put on your safety goggles and ensure chairs are out of the way. You must conduct this experiment standing up.
- 3. Prepare an ice-water bath in the large tray
- 4. Set up the Bunsen burner on the heat mat. Place the tripod and gauze above the Bunsen burner.
- 5. Place the can on top of the gauze. Practise either:
 - a. Using the heat resistant gloves to remove the can, secure the lid and place in the tray.
 - b. Using the tongs to remove the can, then tip it over quickly so that the open end of the can is submerged in the ice-water. If this is not done quickly, the water vapour will condense and pour out, and air will refill the can.
- 6. Place the can on the desk and pour approximately 25 cm³ of water from the beaker into the can.
- 7. Place the can on top of the gauze. Ensure that it is stable.
- 8. Check that you are happy with the set-up and that everything is secure does it match your teacher's demonstration of the set-up?
- 9. Light the Bunsen burner and ensure it is beneath the gauze and the can.
- 10. Observe the can.

Eventually you should notice steam rising from the can.

- 11. When steam is rising consistently from the can, it is time to perform the next stage of the experiment. Turn off the Bunsen burner, and either:
 - a. Use the heat-resistant gloves to remove the can, secure the lid and place in the tray.
 - b. Use the tongs to remove the can, then tip it over quickly so that the open end of the can is submerged in the ice-water.

If this is not done quickly, the water vapour will condense, and air will refill the can.

12. Observe what happens next.

This will happen quickly; make sure you pay close attention.

- 13. Record your observations from the experiment.
- 14. Leave the can in the tray to cool. Leave the Bunsen burner, tripod, gauze and tongs on the heat mat to cool. Return the tongs/gloves, goggles and beaker.

Worksheet E: Observations of the imploding can



Initially the water and air inside the can are at room temperature. can The pressure of the air inside the can is equal to the atmospheric pressure outside the can. Arrows represent the force spread over the area of the can. Note that there are equal amounts inside and outside because the pressure is the same. water The can is placed over the Bunsen burner and is heated. Steam coming out of the top of the can means that The steam now fills the can. This means the pressure in the can ... The can is carefully lifted off the tripod and immersed in the ice-water. The steam inside of the can The can ... This is because the pressure inside the can

Complete the diagrams and explanations to record your observations of the experiment.

Worksheet F: Making observations



Look at the picture below.



Source: http://www.railroad.net/forums/viewtopic.php?f=9&t=63568

Answer the following questions:

- 1. What do you see in this picture? (Observations)
- 2. What do you think about this picture? (Deductions)
- 3. What do you think has led to this? (Hypothesis)
- 4. What are the reasons for your hypothesis? (Scientific thinking)

Worksheet G: Calculating pressure

Carry out the calculations of pressure, force and surface area.

Recall: pressure = $\frac{\text{force}}{\text{area}}$

- 1. A worker has a weight of 800 N and wears big work boots.
 - a. Calculate the worker's pressure when he stands with both feet on the ground. Each boot has a surface area of 0.032 m².
 - b. The worker uses his ladder to clean a very large tanker. The two feet of the ladder have a surface area of just 0.005 m². Calculate the pressure of the ladder on the ground with the worker on it.
- 2. If a person falls into the water of a frozen lake, a ladder can be used to prevent the rescuers also falling through the ice. Explain how a ladder can be used in this way.
- 3. The empty tanker has a weight 210000N. The total surface area of the tankers steel wheels is 0.002 m².
 - a. Calculate the pressure of the tanker on the ground.
 - b. If a lorry with the same weight as the tanker exerts a pressure of 150000Pa, what is the total surface area of its rubber wheels in contact with the ground?
 - c. Would loading the tanker and the lorry cause an increase or decrease of pressure on the ground? Explain each case carefully and consider all factors.
- 4. The tanker has a surface area of 40m². The pressure inside the tanker is 78000 Pa below the atmospheric pressure outside the tanker when it is crushed. Calculate the force that has been applied to the tanker by the atmosphere around it.

Extension:

5. Estimate the force required to crush the can used in the experiment in the last lesson.

Worksheet H: Taking precautions



The tanker pictured on **Worksheet F** became crushed because it was steam-cleaned and then sealed. There were no vents to re-introduce the air that had been displaced. As the steam inside cooled, the pressure inside the tanker decreased to become much lower than the atmospheric pressure. The tanker collapsed under the immense force created by the pressure difference, just like the can in our experiment.

As the owner of a tanker company, write a letter to the staff responsible for cleaning the tankers. Explain the safety issues and precautions that they must take to ensure the tankers do not implode after being cleaned. Include a clear explanation of pressure so that they understand why these precautions need to be taken.

Worksheet I: Match up

Match up the sentences below:

Force divided by area is the equation for	the pressure inside is lower than outside.
For a given force, increasing the area	the pressure inside is higher than outside.
For a given area, increasing the force	reduces the pressure.
The can in our experiment implodes because	in all directions.
An aerosol can might explode when heated because	pressure.
A force may produce	heating.
Atmospheric pressure acts	a change in the size and shape of a body.
Pressure can be increased by	increases the pressure.
Pressure can be decreased by	cooling.



Worksheet A: Answers

- 1. Fill in the spaces below:
 - a. 1 Pascal = <u>1</u> N/m²
 - **b.** For a given area, increasing the force will *increase* the pressure.
 - **c.** For a given force, increasing the area will <u>decrease</u> the pressure.
- 2. Discuss and answer the following questions.
 - a. Why do tractors have large flat tyres?

Large flat tyres have a large surface area. This spreads the weight/force of the tractor, reducing the pressure so it does not get stuck in the mud.

- b. Why do you sink in snow less when you wear skis? Skis have a larger surface area than your feet so this reduces the pressure on the snow even though your weight is identical
- c. Why do camels have large flat feet?Large flat feet have a large surface area.

This spreads the weight/force of the camel, reducing the pressure and helping the camel to walk on sand without sinking.









Worksheet B: Answers

Atmospheric pressure

- is caused by the weight of air above us
- decreases with altitude
- acts in all directions
- Average sea-level pressure is 101 000 Pa.

Write observations for the experiments below:





Worksheet B: Answers



Press the plunger into a smooth surface. Try to	What do you observe?
pull it away.	The plup are is difficult to pull away
	The plunger is allicult to pull away.
	Explain why this happens.
	air inside is pushed out, leaving a lower
	pressure inside the plunger cup than
	outside. The atmospheric pressure acting
	on the outside of the plunger cup holds
	it in place.
The bottle has two holes, one higher than the other. Fill it with water and put on the lid. If the	What do you observe?
	The water streaming out of the lower
top hole is covered, what happens to the water coming out of the lower hole?	hole can be stopped by covering the top
	hole.
	Explain why this happens.
4	The water only streams out of the lower
	hole if air can get into the bottle via the
	top hole. That air exerts pressure on the
	water in the bottle, pushing it out of the hole at the bottom

Worksheet F: Answers

Look at the picture below.



Source: http://www.railroad.net/forums/viewtopic.php?f=9&t=63568

Answer the following questions:

- 1. What do you see in this picture? (Observations)
- 2. What do you think about this picture? (Deductions)
- 3. What do you think has led to this? (Hypothesis)
- 4. What are the reasons for your hypothesis? (Scientific thinking)
- The tanker is compressed or squashed. It looks like it has been sucked in on itself.
- 2. The tanker may have had some structural problems. Something has occurred inside the tanker. This is an issue that could be dangerous or costly to the company
- 3. Something has happened inside the tanker to lower the pressure so that the atmospheric pressure is greater leading to the collapse.
- 4. The bending in the tanker looks like the same effects of the steam condensing in the can in the experiment. It is likely that heating has occurred inside the container and then it has been sealed so that the internal pressure has reduced and the force of the higher atmospheric pressure outside has caused it to implode.

Worksheet G: Answers



- 1. A worker has a weight of 800 N and wears big work boots.
 - a. Calculate the worker's pressure when he stands with both feet on the ground. Each boot has a surface area of $0.032 \, \text{m}^2$.

 $P = F/A = 800 / (2 \times 0.032) = 12500 Pa$

b. The worker uses his ladder to clean a very large tanker. The two feet of the ladder have a surface area of just 0.005 m². Calculate the pressure of the ladder on the ground with the worker on it.

P = F/A = 800 / 0.005 = 160 000 Pa

2. If a person falls through the surface ice into the water of a frozen lake, a ladder can be used to prevent the rescuers also falling through the ice. Explain how a ladder can be used in this way.

It can be laid down flat on the ice. This provides a larger surface area than if a person walked out on the ice. This spreads the weight of the rescuers and reduces the risk of the ice breaking further.

- 3. The empty tanker has a weight 210000N. The total surface area of the tankers steel wheels is 0.002 m².
 - a. Calculate the pressure of the tanker on the ground.

P = F/A = 210000 / 0.002 = 105 million Pa

- b. If a lorry with the same weight exerts a pressure of 150000Pa, what is the total surface area of its rubber wheels in contact with the ground?
 A = F/P = 210000 / 150000 = 1.4 m²
- c. Would loading the tanker and the lorry cause an increase or decrease of pressure on the ground? Explain each case carefully and consider all factors.

Loading the tanker increases the force, which leads to an increase of pressure.

Loading the lorry increases the force, which leads to an increase of pressure.

However, loading the lorry may cause the rubber tyres to be compressed, which

increases the surface area of the tyres in contact with the ground. This would mean

the pressure does not increase as much as for the tanker with the same weight.

4. The tanker has a surface area of 40 m². The pressure inside the tanker is 78000 Pa below the atmospheric pressure outside the tanker when it is crushed. Calculate the force that has been applied to the tanker by the atmosphere around it.

 $F = P \times A = 3120000N$

Extension:

Estimate the force required to crush the can used in the last lesson.
 Learners should use values from the tanker question to scale down the force required.
 The final answer is not important, instead look at learner's scientific thinking and application of logic.

Worksheet I: Answers

Match up the sentences below:

Force divided by area is the equation for	pressure.
For a given force, increasing the area	reduces the pressure.
For a given area, increasing the force	increases the pressure.
The can in our experiment implodes because	the pressure inside is lower than outside.
An aerosol can might explode when heated because	the pressure inside is higher than outside.
A force may produce	a change in the size and shape of a body.
Atmospheric pressure acts	in all directions.
Pressure can be increased by	heating.
Pressure can be decreased by	cooling.



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