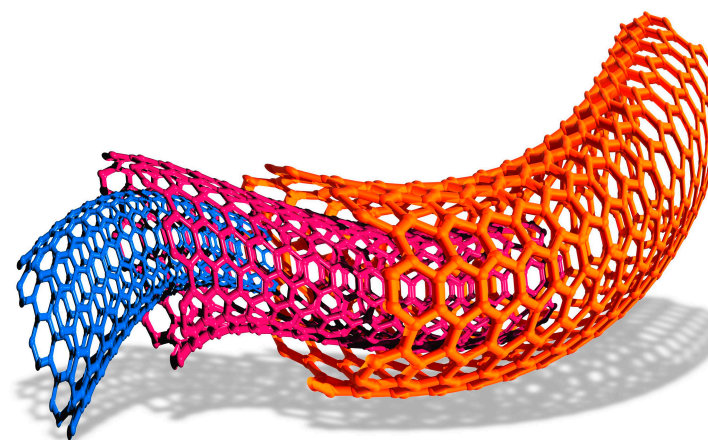


Interactive Example Candidate Responses

Paper 6 (May / June 2016), Question 1

Cambridge IGCSE™
Chemistry 0620



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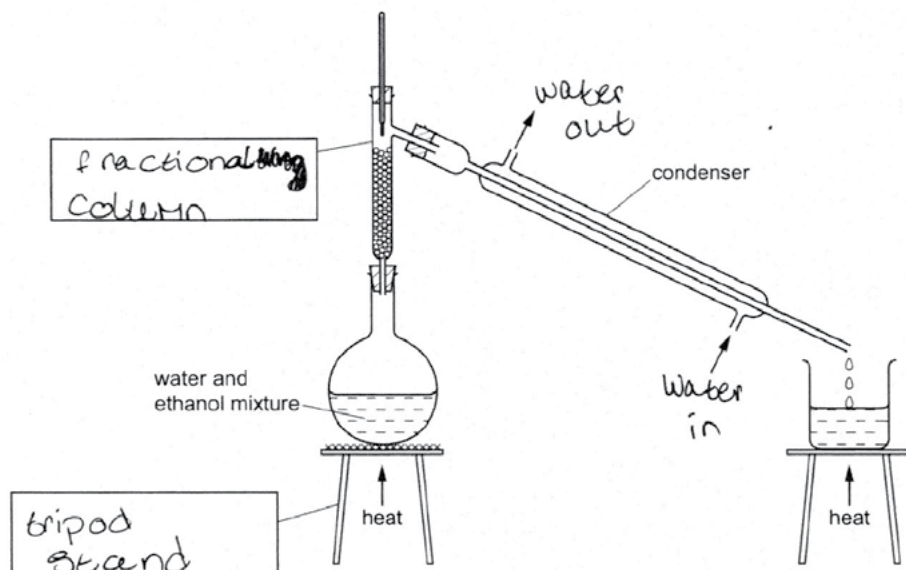
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- 1 The diagram shows the apparatus used to separate a mixture of water, boiling point 100°C , and ethanol, boiling point 78°C .



(a) Complete the boxes to name the apparatus. [2]

(b) Label the arrows on the condenser. [1]

(c) Identify **one** mistake in the apparatus. applying heat to beaker [1]

(d) Which liquid would collect first? Explain your answer. ethanol; its boiling point is lower than water's boiling point [2]

(e) Why would it be better to use an electrical heater instead of a Bunsen burner to heat the water and ethanol mixture? ethanol is flammable [1]

[Total: 7]

Select page

Your Mark

1(a)

1(b)

1(c)

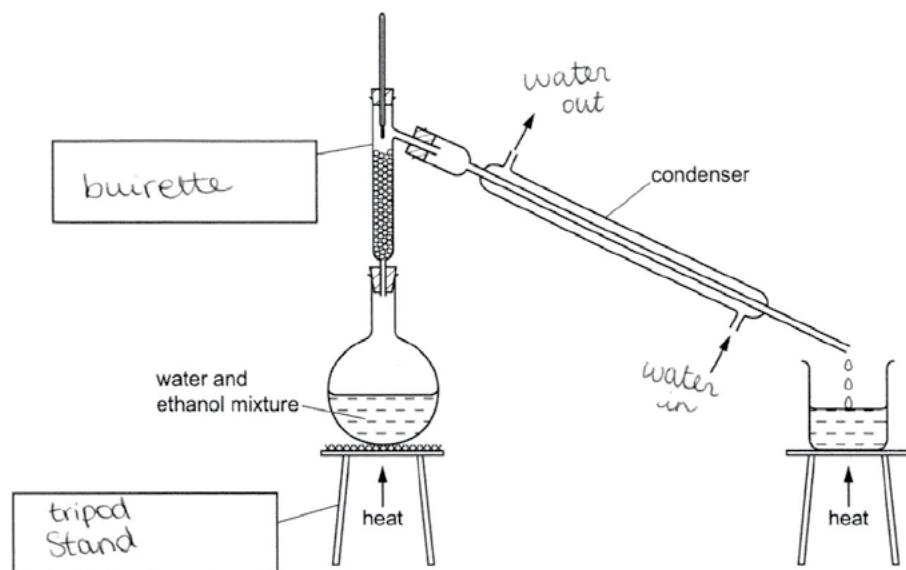
1(d)

1(e)

Q1 Mark scheme

(a)	fractionating column; tripod;
(b)	<u>water</u> labelled twice;
(c)	heat under (the collecting) beaker;
(d)	M1 ethanol; M2 lowest/lower boiling point;
(e)	ethanol is flammable;

- 1 The diagram shows the apparatus used to separate a mixture of water, boiling point 100°C , and ethanol, boiling point 78°C .



- (a) Complete the boxes to name the apparatus. [2]
- (b) Label the arrows on the condenser. [1]
- (c) Identify **one** mistake in the apparatus.
 heat applied to condensed liquid [1]
- (d) Which liquid would collect first? Explain your answer.
 ethanol, it has a lower boiling point [2]
- (e) Why would it be better to use an electrical heater instead of a Bunsen burner to heat the water and ethanol mixture?
 you can choose the exact temperature [1]

[Total: 7]

Select
page

Your
Mark

1(a)

1(b)

1(c)

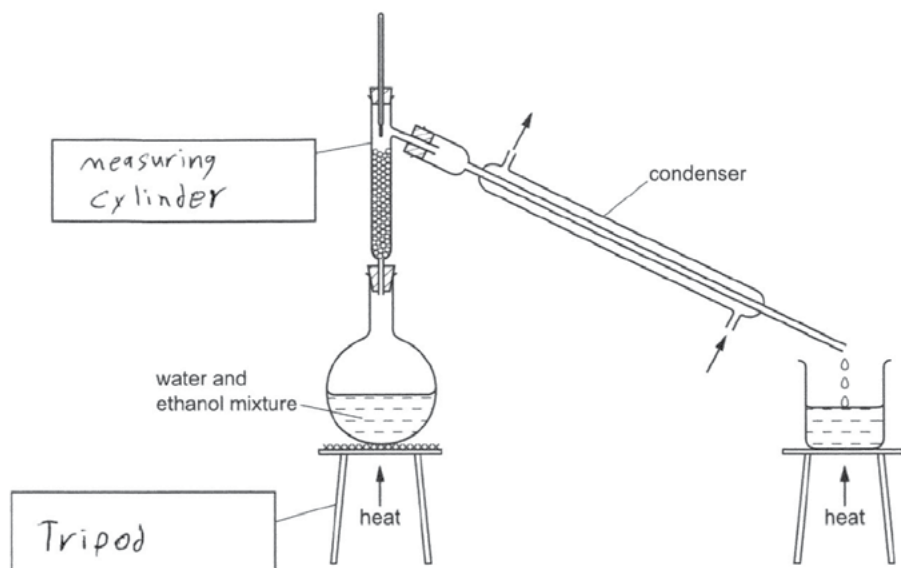
1(d)

1(e)

Q1 Mark scheme

(a)	fractionating column; tripod;
(b)	<u>water</u> labelled twice;
(c)	heat under (the collecting) beaker;
(d)	M1 ethanol; M2 lowest/lower boiling point;
(e)	ethanol is flammable;

- 1 The diagram shows the apparatus used to separate a mixture of water, boiling point 100°C , and ethanol, boiling point 78°C .



(a) Complete the boxes to name the apparatus. [2]

(b) Label the arrows on the condenser. [1]

(c) Identify **one** mistake in the apparatus.
~~(Water and ethanol mixture)~~ ^{Heat applied on burrette.} ~~(Heat in the)~~ ^{collecting} [1]

(d) Which liquid would collect first? Explain your answer.
 Water, because it will get seperated from ethanol. [2]

(e) Why would it be better to use an electrical heater instead of a Bunsen burner to heat the water and ethanol mixture?
 For accurate heating. [1]

[Total: 7]

Your
Mark

1(a)

1(b)

1(c)

1(d)

1(e)

Q1 Mark scheme

(a)	fractionating column; tripod;
(b)	<u>water</u> labelled twice;
(c)	heat under (the collecting) beaker;
(d)	M1 ethanol; M2 lowest/lower boiling point;
(e)	ethanol is flammable;

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